# The High School–University Transition in Chemistry

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### **Chemical Education Survey:**

- Pilot study in 2006-7
- 1st major survey in 2007-8
- 2<sup>nd</sup> major survey in 2008-9
- Mixed qualitative/quantitative study

What factors contribute to a successful high school—university transition?

What can schools and universities do to help students manage this transition?

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### The Survey Cohort:

- CHM 138F (Introduction to Organic Chem.)
- CHM 139F (General Physical Chemistry)
- CHM 151Y (Advanced Introductory Chem.)

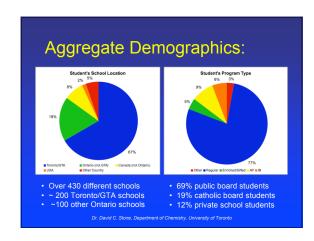
Year	Enrolment	Surveys	Response
2006-7	1830	320	17.5%
2007-8	1803	536	29.3%
2008-9	1723	414	24.0%
Total:	5356	1270	23.7%

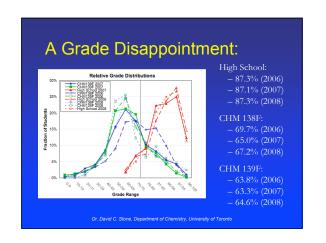
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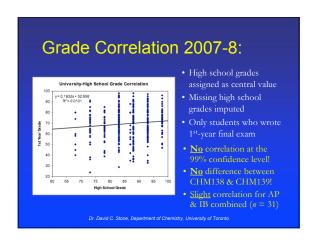
### Who Are Our Students?

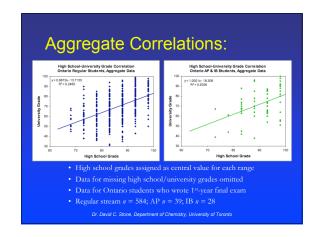
Category	2006-7	2007-8	2008-9
Female <sup>1</sup> :	-	60.6%	59.4%
Male <sup>1</sup> :	-	39.4%	40.6%
Toronto/GTA:	-	68.9%	69.1%
Total Ontario:	86.4%	84.4%	84.5%
Regular stream:	68.1%	82.3%	78.8%
Semestered:	-	58.4%	65.1%
Native English-speaker:	-	44.8%	45.9%*
Independent Study:	56.0%	57.7%	44.9%

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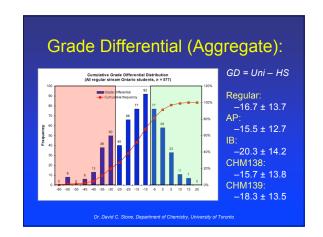




### **Aggregate Correlation Data:**

· Ontario students with full grades reported

Category	n	r	t	p (t)
All	690	0.490	14.8	< 0.0001
CHM138F	489	0.458	11.4	0.00012
CHM139F	201	0.580	10.05	< 0.0001
Regular	584	0.496	13.8	< 0.0001
AP/IB	67	0.454	4.102	< 0.0001



### Related US Study:

Survey of 12 US colleges & universities:

- R. H. Tai, P. M. Sadler, and J. F. Loehr
   J. Res. Sci. Teaching, 2005, 42(9), 987-1012
- R. H. Tai, R. B. Ward, and P. M. Sadler
   J. Chem. Ed., 2006, 83(11), 1703-1711
- R. H. Tai and P. M. Sadler

   J. Chem. Ed., 2007, 84(6), 1040-1046

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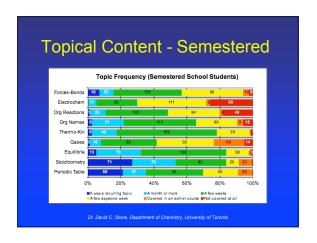
### **Top Grade Predictors:**

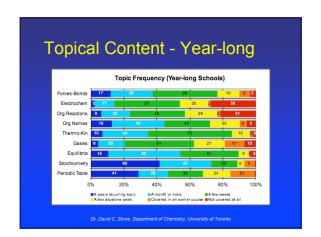
- Last HS Math Grade (AP and/or calculus) SAT Math score also highly significant
- 2. Last HS science grade (not specifically chemistry)
- 3. Time spent on stoichiometry (recurring topic)
- 4. AP instead of regular chemistry; emphasis on understanding *vs.* memorization

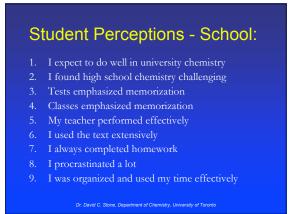
Tai and Sadle, op. cit.

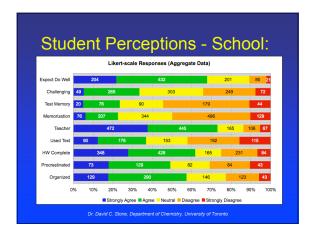
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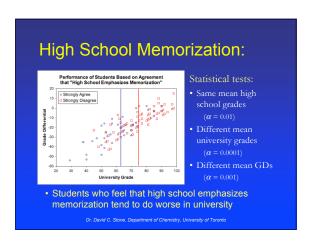
# Topical Content: Atoms & periodic table (electron config., periodicity, etc.) Stoichiometry (chemical reactions & equations) Equilibria (reactions, acid/base, solubility) Gases (properties, gas laws) Thermodynamics & Kinetics (energy, Hess' Law, etc.) Organic Chemistry (naming, groups, structure) Organic Chemistry (reactions, mechanisms) Electrochemistry (redox, galvanic & voltaic cells) Forces & Bonding (VSEPR, van der Waal's, etc.) Ontario Curriculum: Grade 11 and Grade 12 (2000-9)







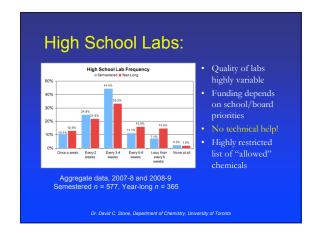




### **Focus Group Themes:**

- Teaching & evaluation practices
- Use of text (by student or teacher)
- Self-directed learning & pace of material
- Relevance & complexity of labs
- Organic coverage from curriculum

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### **Diagnostics - Content:**

"The decline in A and B grades has been accompanied by a marked increase in F and dropped grades." Nelson Hovey & Albertine Krohn, ICE 1958 (35) 507-509

- California Chemistry Diagnostic Test
  - ACS Examinations Institute

Arlene Russell, JCE 1994 (71) 314-317

- CIC Chemistry Exam (Part A)
  - based on Pan-Canadian Protocol, Grade 12

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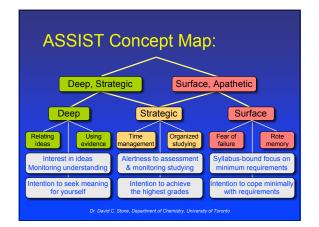
### Diagnostics - Style/Skills:

"The most accurate predictive measure of degree results is generally first-year grades, but the highest proportion of failure occurs during the first year."

Approaches & Study Skills Inventory for Students (ASSIST)

Deep	Surface	
Strategic	Apathetic	

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- U of T 1st-year students, for participating
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